A new validated high-performance liquid chromatography method for standardization of rosmarinic acid in *Salvia* extracts

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**Abstract**

**Introduction:**

Tea bags or infuses of *Salvia* species from Lamiaceae family are traditionally used for the treatment of bronchitis, cough, and throat inflammations (1). They are known for antioxidant properties mainly related to the presence of rosmarinic acid (RA). Therefore it is necessary to develop a reliable analytical method for RA assay for standardization of *Salvia* species and also other plants containing RA like *Melissa*, *Origanum*, *Lavandula*, *Rosmarinus*, *Thymus*, *Zataria*, *Mentha*, *Perovskia*, *Zhumeria*, and *Satureja* species. In this study using a suitable extraction method by removing unwanted components present in crude methanol extract, phenolic content containing RA was extracted from dry powders of six *Salvia* species. Then, a suitable high-performance liquid chromatographic (HPLC) method was optimized for quantification of RA in *Salvia* species.

**Methods:**

HPLC analysis was done on a Waters system, equipped with 515 HPLC pump and waters 2487 dual wavelength absorbance detector. The column was a Nova-Pak C18 (3.9 × 150 mm), and Millenium software was used for the determination of the compounds and processing the data. The method was validated according to USP 32 requirements.

**Results:**

Among the investigated 6 species, *S. virgata* was the richest in RA level, demonstrating 3.50 ± 0.12 mg/g, followed by *S. sclarea* and *S. chloroleuca* showing 1.65 ± 0.08 and 1.65 ± 0.21 mg/g. *S. ceratophylla* with 0.10 ± 0.01 mg/g of RA in dried plant powder was the poorest.

**Conclusion:**

The validated HPLC method allows determination of amounts as low as 2.5 µg/mL of RA and linearity in the ranges of 2.5-25 µg/mL and 100–600 µg/mL, which is suitable for standardization of *Salvia* species in traditional and pharmaceutical formulations.
Materials and Methods

Collection and authentication of the plants

The aerial parts of Salvia species (Lamiaceae) were supplied from the Khorasan, Iran and was identified by Dr. L. Ghaem-maghami, Department of biology, Faculty of Science, University of Isfahan. Herbarium specimens of S. ceratophylla with voucher number: 36207, S. chloroleuca No. 36528, S. macrosiphon No. 43000, S. nemorosa No. 40912, S. scarea No. 32961, and S. virgata No. 43965 were deposited in the herbarium center of the Faculty of Pharmacy, Mashhad University of Sciences (Iran).

Instrumentation

HPLC (High-performance liquid chromatographic) analysis was done on 515 HPLC pump, waters 2487 dual wavelength absorbance detector (Waters, Milford, MA, USA). The column was a Nova-Pak C18, 3.9 × 150 mm (Waters, Milford, MA, USA) and Millenium software was used for the determination of compounds and processing the data.

Chemicals

Acetonitrile HPLC-grade solvent was purchased from Caledon Company (Canada). RA, as analytical standard, was purchased from Sigma-Aldrich Corporation, USA.

Sample preparation and cleanup procedure for analysis

Aerial parts of each Salvia species were dried at room temperature in the dark, separately. 1 g of crushed dried powder was macerated in 10 mL of ethanol: water: NaHCO3 (350:150) as solution (A) and CH3CN:H2O (54.5:445:0.5) as solution (B) with a stepwise gradient system with a flow rate of 0.8 ml/min while the separation time was 14 minutes and the injection volume was 5 µL. For quantitative analysis, the wavelength was set at 328 nm and data was collected by Millenium Chromatography Software (Build 1154, Waters Corporation, Milford, USA).

Selectivity

The selectivity was confirmed by HPLC analysis of RA in Salvia extracts. The ability of method to separate closely ingredients from RA band was calculated by assessing resolution, tailing factor and theoretical plate number. Resolution (Rs) is a function of theoretical number of plates (N), separation factor (α), and the capacity factor (k). Identification of RA in chromatograms was done by using the following formulas:

Resolution (Rs) = \frac{N_{theoretical}}{2} = \frac{\alpha - 1}{\ln(\alpha)}

Selectivity (Se) = \frac{1}{\alpha} + 1

Capacity factor (k) = \frac{t_{R} - t_{0}}{t_{0}}

where:

- \(t_{R}\) is the retention time of the analyte,
- \(t_{0}\) is the retention time of the solvent front.

The selection of wavelength for HPLC analysis was done in the range 210–330 nm. A good selectivity depends highly on the molar absorption of the analyte to be high enough for detection of low concentrations of RA in its working range. It depends also on the UV absorption of other ingredients especially those with retention time beyond the RA and mobile phase UV cut off point. UV spectrum of RA showed that λmax absorption was 196, 221, 291, and 328 nm. Among them, λmax of 328 with more molar absorption and better mobile phase (acetonitrile: water) UV cut off point, and less overlap with other bands in HPLC chromatogram was selected.

Determination of suitable wavelength

Since the HPLC analysis was based on UV absorption and because many other phenolics or aromatic diterpenes in Salvia species show UV spectra with λmax in the range of 210–330 nm, a good selectivity depends highly on the molar absorption to be high enough for detection of low quantities of RA in its working range. It depends also on UV absorption of other ingredients especially those with retention time beyond the RA and mobile phase UV cut off point.

Preparation of the Standard Solutions

1000 µg analytical standards were diluted in 1 mL of HPLC-grade acetonitrile: water (1:1) to prepare the stock solution of 1000 µg/mL. The calibration method was prepared by serially diluting the stock solution to concentrations of 600, 500, 400, 200, and 100 µg/mL.

Instrumentation and chromatographic conditions

RA was determined through an external standard calibration method on a Waters HPLC 515 system, equipped with 2487 dual wavelength UV detector. External standard calibration is a simple method suitable for reproducible dilution volumes and injection volumes and if the recovery of an analyte is about 100%, it can be expected to be similar for all expected samples. The chromatographic analysis was done on a reverse phase Nova-Pak C18, 3.9 × 150 mm column (Waters, Milford, MA, USA). Temperature was maintained at 45ºC. The mobile phase was consisted of CH3CN:H2O:H3PO4 (545:445:0.5) as solution (A) and CH3CN:H2O (350:150) as solution (B) with a stepwise gradient system with a flow rate of 0.8 ml/min while the separation time was 14 minutes and the injection volume was 5 µL. For quantitative analysis, the wavelength was set at 328 nm and data was collected by Millenium Chromatography Software (Build 1154, Waters Corporation, Milford, USA).

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Standardization of rosmarinic acid in Salvia extracts
by spiking the sample with standard RA. N and Rs were respectively calculated by using equations, N=5.54 (Rt / W0.5)², and Rs=2(Rt – Rs)/(Wp + Ws), where Rt, Rs, and Wp are retention times and band width of RA, Wp, is peak width at half height of RA peak, and Rs, and We are related to closed eluted peak to RA peak in chromatogram (9,10). Tailing is a condition in which the target peak is skewed after the peak apex. It happens especially in gradient elution systems and where the mobile phase and stationary phase concentrations of solute are not in equilibrium especially in fast mobile phase flow or where there is a slow kinetic desorption from the stationary phase. Tailing factor was defined according to the US Pharmacopoeia through equation, T0.05 = W/2*W0.05

Limits of detection and quantification
The minimum amount of analyte which can be detected reliably is often referred to LOD (limit of detection) and is a peak which its signal-to-noise ratio is at least 3:1. The minimum quantifiable amount (LOQ) is the concentration which can be quantified with a specified level of accuracy. LOQ was calculated by injecting different concentrations of standard with signal-to-noise ratios of 10, 15, 20, and 25 and calculating LOQ with certain level of precision with relative standard deviation (RSD) less than 6, experimentally (9,11).

Repeatability and intermediate precision
Precision is a multivariate factor. It could be controlled by minimizing different sources of errors in quantification. One of these is integration error especially in unresolved peaks (Rs <1.5) and in low standard concentrations (range beyond the LOQ). Other sources of error are injection error and instrument error which should be controlled by repeatability test. It was carried out by a simple intra-day injection method by the same analyst through three repetitive injections of the standards within the selected range in one day (11). There are also additional sources of errors like environmental errors and long-term variability that intermediate in analysis. It was calculated by inter-day precision method through quantification of three replicates of standards in the selected range in three consecutive days in a week in the same lab and same analyst (n = 9). For a precise and suitable method RSD less than 3 is required (12).

Accuracy
One of the methods to be assured from the accuracy of HPLC quantification method is recovery tests. It ensures that the real amount of component is measured. It was performed by adding the RA as standard beyond the amount in the real sample in 1 g of a blank powder. It was extracted as was mentioned in the sample preparation method. Then evaporated sample was solubilized into 6 mL of HPLC mobile phase solvent. Three determinations were carried out and the recovery percentage was calculated in every case (11).

Results
The HPLC method carried out in this study was aimed to optimize the resolution (Rs), reduce runtime, tailing factor, and limit of quantification of RA in Salvia species.

Optimizing the extraction method
Two extraction methods consisted of crude methanol extract and special method for extraction of phenolic components were compared to check the effect of the procedure on the extraction. Finally, for removing unwanted components present in crude methanol extract, and to increase resolution and separating factor, a special method was designed for removing unwanted materials. Phenolic content containing RA was extracted from dry powder in ethanol: water: NaHCO3 (65:30:5) to produce sodium rosmarinate which is a more water-soluble ionizable conjugate and improves the solubility and extraction rate. After filtration and concentration in vacuum for removing sugars, glycosides, and unwanted polar components present in residue, and to prevent additional interferences, phenolic content was back extracted using liquid extraction in a separating funnel by partitioning between dilute HCl and diethyl ether. In partitioning, when RA is in the ionized form (pH = 9), it is more soluble in the aqueous phase. Likewise, when it is in protonated uncharged state (pH = 1), it can be partitioned and transferred into the organic layer (diethyl ether phase). However in this study, for a good and more precise back extraction, using HPLC analysis, by selection of different organic solvents (diethyl ether, ethyl acetate, chloroform), and changing pH (7, 4.5, 3, 1), by selection of diethyl ether as solvent and controlling the pH to 1, the selectivity and recovery was checked and optimized.

Method development
For separation of RA, a conventional USP L1 column, Waters Nova-Pak C18: 3.9 × 150 mm (4 μm), was selected. Then, using gradient method from 0% to 100% acetonitrile in water, the solvent composition suitable for RA elution from 10% to 70% was selected. Then, to improve the retention of RA as a weak acid, the pH of the mobile phase was controlled. The mobile phase pH could change the ionizable species of RA in solution which affects its retention and selectivity. It changes ionization of both RA and residual silanol groups of column. Reducing the pH to 2.8 by adding phosphoric acid (1%) as modifier made both RA and silanol groups protonated and improved selectivity and peak tailing. Then, the column temperature was enhanced and kept constant to 40°C to reduce solvent viscosity, back pressure, and variation of retention time of analyte. Finally, the mobile phase was consisted of CH3CN:H2O:H3PO4 (54.5: 445: 0.5) as solution (A) and CH3CN:H2O:H3PO4 (345.5:150: 0.5) as solution
Demonstrate HPLC chromatogram of sage (B). It was started with A: B (100:0) hold for 2 minutes, then 0–100% B in 12 minutes, with equilibration with 0%–100% A for 4 minutes. The flow rate was 1 mL/min and temperature was kept constant at 40°C. UV detector was set to single wavelength mode at 328 nm with a time constant of 0.1 seconds. Small injection volume of 5 µL was injected to increase resolution.

Determination of rosmarinic acid
Using high pressure liquid chromatography, RA was determined using external standard calibration method. The retention time was observed to be 9.1 minutes (Figure 1).

Linearity and range
Using USP 35-NF 30 definition of signal to noise ratio, the LOD and LOQ were calculated as 0.8 µg/mL and 2.5 µg/mL for RA. Sensitivity expressing as the LOD and LOQ values could be reduced more in the case of using isocratic mobile phase system, or increasing the injection size, but the current gradient mobile phase system possessed better selectivity and resolution. As clear in Table 1, using the millennium processing software, the calibration curve was determined by linear regression in two different working ranges, the range of 2.5-25 µg/mL and the range of 100-600 µg/mL (Figure 2).

Precision
The instrumental precision by intra-day results and intermediate precision through inter-day results of repetitive RA quantitated in different concentrations in the selected linear range, 100 to 600 µg/mL, as are presented in Table 2.

Accuracy
Accuracy was evaluated as recovery test after spiking of 200 µg of RA standard to 1 g of the Myrtus communis powder as blank powder. It was extracted as mentioned above in sample preparation. Three determinations were done and the recovery percentage was found to be 92.4 ± 4.36%.

Rosmarinic acid assay in sage samples
Figure 3 demonstrate HPLC chromatogram of sage samples and Table 3 reports the amounts of RAs in six sage samples collected from North East of Iran. Each value was the mean of three extractions. The content of RA ranged from 0.1 to 3.2 mg/g of dried powder.

Discussion
RA is found in some of Labiateae family plants like Salvia, Melissa, Origanum, Lavandula, Rosmarinus, Thymus, Zataria, Mentha, Perovskia, Zhumeria, and Satureja species (13). It is also found in species of the Boraginaceae and in some fern and hornwort species (14). The major problem in their use is the plant-to-plant variability of RA. Therefore, using a validated chromatographic method, RA as one of their main active ingredients should be assayed and standardized. Another problem was chlorophylls, glycosides, and unwanted secondary metabolites.

Figure 1. HPLC chromatogram of rosmarinic acid (100 µg/mL) at 328 nm.

Table 1. Linearity, limit of detection and minimum quantifiable amount parameters of rosmarinic acid standard

<table>
<thead>
<tr>
<th>Compound</th>
<th>LOD µg/mL</th>
<th>LOQ µg/mL</th>
<th>R² square</th>
<th>Equation</th>
<th>Linear range (µg/mL)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Rosmarinic acid</td>
<td>0.8</td>
<td>2.5</td>
<td>0.999</td>
<td>y = 26469x + 1E+06</td>
<td>100-600</td>
</tr>
<tr>
<td>Rosmarinic acid</td>
<td>0.8</td>
<td>2.5</td>
<td>0.997</td>
<td>y = 26250x + 14435</td>
<td>2.5-25</td>
</tr>
</tbody>
</table>

Abbreviations: LOD, limit of detection; LOQ, limit of quantitation.
with overlapped peaks and interferences in HPLC chromatogram. Therefore, using a suitable extraction method by removing unwanted components present in crude methanol extract, phenolic content containing RA was extracted from dry powder. Then, a suitable HPLC method was optimized. It allowed determination of amounts as low as 2.5 µg/mL of RA and linearity in the ranges of 2.5-25 µg/mL and 100–600 µg/mL. Finally, RA levels of 6 *Salvia* species grows wild in North-East of Iran were determined via HPLC validated method. The obtained results were summarized in Table 3. Among the investigated 6 species, *S. virgata* was the richest in RA level, demonstrating 3.50 ± 0.12 mg/g, followed by *S. sclarea* and *S. chloroleuca* showing 1.65 ± 0.08 and 1.65 ± 0.21 mg/g, while *S. ceratophylla* with 0.10 ± 0.01 mg/g of RA in dried plant powder was the poorest.

In this regard, Wang et al developed an HPLC method for the determination of rosemarinic in rosemary, sage, thyme, spearmint, balm, and lavender. They injected directly crude methanol extract as sample using a gradient elution consisted of methanol: water: phosphoric acid. It was a simple and suitable method but with a long Run time of 25 minutes and a crowded chromatogram (15). In another study Bandoniene et al (16) developed LC-Mass chromatographic method for detection of RA in the leaves of *Salvia* species including *S. officinalis*, *S. glutinosa*, *S. aethiopis*, *S. sclarea*, and *Borago officinalis*. They used an RP-18 column (250 × 3 mm, 5 µm) with a gradient system of water: acetic acid (98:2) and acetonitrile at a flow of 0.3 mL/min at 280 nm. It was a fast and simple LC-PDA-Mas tandem method but with long run time of 30 minutes. Calibration curve was calculated over a wide working range of 25–500 µg/mL and higher LOQ of 6 µg/mL. Unfortunately, this method was not validated in that paper. In another study by Lopez-Arnaldos et al in 1995, they used a spectrophotometric method to directly determine RA in unpurified methanol extracts by a complexation reaction with Fe²⁺ (17). It was a simple and fast chemical method but could not differentiate between RA and other caffeic acid derivatives closely related to RA and a more specific chromatographic method was preferred. Bonoli et al also reported a suitable fast method for determination of both carnosic and RAs in rosemary species by capillary zone electrophoresis with acceptable run time of 10 minutes and good selectivity (18). The main problem in HPLC quantification of RA is complexity of the total extract of *Salvia* species which contains a variety of different phytochemicals that might interact with chromatogram peak of RA. On the other hand, RA in neutral pH solvent is ionized and attached to column silanol groups which lead to widening and tailing. RA has an ionizable structure which causes particular

![Figure 2. Calibration curve of Rosmarinic acid using HPLC method and acetonitrile: water as mobile phase with pH adjusted by adding H₃PO₄ to 2.8 at 328 nm. Using the millennium processing software, the calibration curve was determined by linear regression in the range of 100-600 µg/mL.](image)

### Table 2. Repeatability and intermediate precision assays of standards

<table>
<thead>
<tr>
<th>Concentrations (µg/mL)</th>
<th>Intra-day (n = 9)</th>
<th>RSD%</th>
<th>Intra-day (n = 9)</th>
<th>RSD%</th>
</tr>
</thead>
<tbody>
<tr>
<td>100</td>
<td>3693023 ± 149175</td>
<td>4.0</td>
<td>3201392 ± 195482</td>
<td>6.1</td>
</tr>
<tr>
<td>200</td>
<td>6716530 ± 152209</td>
<td>2.3</td>
<td>6723784 ± 350657</td>
<td>5.2</td>
</tr>
<tr>
<td>400</td>
<td>11756892 ± 786839</td>
<td>6.7</td>
<td>11355005 ± 338960</td>
<td>3.0</td>
</tr>
<tr>
<td>600</td>
<td>17064958 ± 1033107</td>
<td>6.1</td>
<td>16846497 ± 810957</td>
<td>4.8</td>
</tr>
</tbody>
</table>

Abbreviation: RSD, relative standard deviation.
Standardization of rosmarinic acid in Salvia extracts

separation challenges. The pH of mobile phase could be changed to influence the charge state of its ionized form in solution.

When the pH is equal to the pKa the RA rapidly converts between the ionized and non-ionized form and leads to a wide and asymmetric peak shape in HPLC. In fact, two forms of ionized and unionized species have different retention times. The ionized form with more polarity has smaller retention time in reversed phase HPLC. Therefore, in this study the effects of pH on mobile phase, flow, column temperature, and mobile phase composition on the column resolution, tailing factor, and retention time were optimized and a fast, easy and suitable validated HPLC method for RA assay in Salvia extracts or their pharmaceutical dosage forms were designed.

In sum, this validated HPLC method allows determination of amounts as low as 2.5 µg/mL of RA and linearity in the ranges of 2.5-25 µg/mL and 100–600 µg/mL, which is suitable for standardization of Salvia species in traditional and pharmaceutical formulations.

Acknowledgements
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Authors’ contributions
MG coordinated the study and participated in most of the experiments. PM contributed to data analysis and writing and finalizing the manuscript. NS participated in most of the experiments. MM contributed to finalizing the manuscript. All authors read and confirmed final version and publication of the manuscript.

Conflict of interests
The authors have no conflicts of interest.

Table 3. Rosmarinic acid assay (mg/g) in dried aerial parts of 6 sage species grows wild in North East of Iran

<table>
<thead>
<tr>
<th>Salvia species</th>
<th>Voucher No.</th>
<th>Rosmarinic acid (mg/g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>S. ceratophylla</td>
<td>36207</td>
<td>0.10 ± 0.01</td>
</tr>
<tr>
<td>S. chloroleuca</td>
<td>36528</td>
<td>1.65 ± 0.21</td>
</tr>
<tr>
<td>S. macrosiphon</td>
<td>43000</td>
<td>0.58 ± 0.04</td>
</tr>
<tr>
<td>S. nemorosa</td>
<td>40912</td>
<td>0.17 ± 0.01</td>
</tr>
<tr>
<td>S. sclarea</td>
<td>32961</td>
<td>1.65 ± 0.08</td>
</tr>
<tr>
<td>S. virgata</td>
<td>43965</td>
<td>3.50 ± 0.12</td>
</tr>
</tbody>
</table>

Figure 3. High performance liquid chromatography chromatograms of (a) Salvia virgata, (b) S. nemorosa, (c) S. ceratophylla, (d) S. macrosiphon, (e) S. sclarea, (f) S. chloroleuca.
**Ethical considerations**

Ethical consideration has been completely observed by the authors.

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**References**


